

Anexos

Anexo 1

TABLE 7
Hoy Group Molar Attraction Constants at 25°C

Group, z	Bond type	$F/J^{1/2}$ cm ^{3/2} mol ⁻¹	$F_p/J^{1/2}$ cm ^{3/2} mol ⁻¹	V' cm ³ mol ⁻¹
Base value		276.3	—	—
-CH ₃	Saturated	303.4	0.0	21.548
-CH ₂ -	Saturated	269.0	0.0	15.553
>CH-	Saturated	175.9	0.0	9.557
>C<	Saturated	65.5	0.0	3.562
CH ₂ =	Alkene	258.8	66.9	19.173
-CH=	Alkene	248.5	59.5	13.178
>C=	Alkene	172.8	63.0	7.183
-CH=	Aromatic	239.9	62.2	13.417
>C=	Aromatic	200.7	64.8	7.422
-O-	Ether	235.2	216.0	6.462
-O-	Acetal	236.3	10.2	6.462
-O-	Epoxide	360.4	155.9	6.462
-COO-	Ester	668.1	524.1	23.728
>C=O	Ketone	538.0	525.7	17.265
-CHO	Aldehyde	598.6	531.6	23.261
-(CO) ₂ O	Anhydride	1160.4	1159.8	40.993
-COOH	Acid	564.8	415.6	26.102
-OH→	H-bond OH	485.8	485.8	10.647
-OH	Primary (not H-bonded)	673.8	673.8	12.457
-OH	Secondary	591.6	591.6	12.457
-OH	Tertiary	798.6	798.6	12.457
-OH	Phenolic	349.8	349.8	12.457
-NH ₂	Amino, primary	463.5	463.5	17.012
-NH-	Amino, secondary	368.2	368.2	11.017
>N-	Amino, tertiary	125.0	125.0	12.569
-C≡N	Nitrile	725.3	724.5	23.066
-NCO	Isocyanate	733.7	8.2	25.907
HCON<	Formamide	1017.0	724.1	35.830
-CONH-	Amide	1134.6	893.9	28.302
-CONH ₂	Amide	1206.6	989.2	34.297
OCONH	Urethane	1261.9	892.2	34.784
-S-	Thioether	428.3	428.3	18.044
Cl	Primary	419.5	306.8	19.504
Cl	Secondary	426.1	315.0	19.504
Cl ₂	Twinned	701.0	562.5	39.008
Cl	Aromatic	329.3	81.4	19.504
Br	Primary	527.5	122.7	25.305
Br	Aromatic	420.6	100.2	25.305
F	Primary	84.5	73.2	11.200
Conjugation		47.6	-19.8	—
Cis		-14.6	-14.5	—
Trans		-27.6	-27.6	—
4-Membered ring		159.1	200.5	—
5-Membered ring		42.9	84.9	—
6-Membered ring		-48.0	61.0	—
7-Membered ring		92.3	0.0	—
Bicycloheptane ring		46.2	—	—
Tricyclodecane ring		127.8	—	—
Ortho substitution		19.8	-13.3	—
Meta substitution		13.5	-24.3	—
Para substitution		82.4	-33.8	—

Adapted from Hoy, K. L., *The Hoy Tables of Solubility Parameters*, Union Carbide Corporation, 1975 and 1985.
J. Paint Technol., 42, 76, 1970.

Tabla tomada de (Hansen, 2007)

Anexo 2

TABLE 13
Dispersion and Polar Group Molar Attraction Contributions

Structural group	${}^1F_d/J^{1/2} \text{ cm}^{3/2} \text{ mol}^{-1}$		${}^1F_p/J^{1/2} \text{ cm}^{3/2} \text{ mol}^{-1}$ van Krevelen ¹⁵	$({}^1F_p + {}^1V {}^1\delta_p)/J^{1/2} \text{ cm}^{3/2} \text{ mol}^{-1}$ Hansen and Beerbower ¹⁶
	Koenhen and Smolders ¹⁴	van Krevelen ¹⁵		
-CH ₃	411	420	0	—
-CH ₂ -	284	270	0	—
>C-	104	80	0	—
>C<	—	-70	0	—
=CH ₂	—	400	0	—
=CH-	—	200	0	—
=C<	—	70	0	—
	—	1620	0	—
	1640	—	—	—
	1510	1430	110	—
 (o, m, p)	1350	1270	110	—
-F	—	(220)	—	460 ± 50
-Cl	—	450	550	610 ± 200
>Cl ₂	—	—	—	360 ± 50
-Br	—	(550)	—	610 ± 50
-I	—	—	—	665 ± 50
-CN	446	430	1100	1070 ± 100
-OH	202	210	500	510 ± 60
(-OH) _n	—	—	—	n(350 ± 50)
-O-	—	100	400	410 ± 100
-COH	—	470	800	—
-C=O	325	—	—	—
-CO-	—	290	770	800 ± 30
-COOH	409	530	420	450 ± 20
-COO-	395	390	490	510 ± 50
HCOO-	—	530	—	—
-NH ₂	—	280	—	610 ± 200
-NH-	143	160	210	205 ± 30
-N<	—	20	800	—
-NO ₂	440	500	1070	1020 ± 100
-S-	—	440	—	—
=PO ₄ ⁻	—	740	1890	—
Ring	—	190	—	—
One plane of symmetry	—	—	0.50 ×	—
Two planes of symmetry	—	—	0.25 ×	—
More planes of symmetry	—	—	0 ×	—

Tabla tomada de (Hansen, 2007)

TABLE 14
Hydrogen Bonding Parameter Group Contributions

Structural group, z	${}^{\circ}U_z/\text{J mol}^{-1}$ van Krevelen ¹⁵	$-{}^{\circ}U_z = -V^{\circ}\delta_z^2/\text{J mol}^{-1}$ Hansen and Beerbower ⁷⁶	
		Aliphatic	Aromatic
-CH ₃	0	—	—
-CH ₂ -	0	—	—
>CH-	0	—	—
>C<	0	—	—
=CH ₂	0	—	—
=CH-	0	—	—
=C<	0	—	—
	0	—	—
	0	—	—
 (o, m, p)	0	—	—
-F	—	~0	~0
-Cl	400	400 ± 80	400 ± 80
>Cl ₂	—	690 ± 40	750 ± 40
-Br	—	2100 ± 400	2100 ± 400
-I	—	4000 ± 800	—
-CN	2500	2100 ± 800	2300 ± 800
-OH	20000	19500 ± 1700	19500 ± 2100
-(OH) _n	—	n(19500 ± 1700)	n(19500 ± 1700)
-O-	3000	4800 ± 1200	5200 ± 1200
-COH	4500	—	—
-CO-	2000	3300 ± 600	3300 ± 600
-COOH	10000	11500 ± 1000	9400 ± 1000
-COO-	7000	5200 ± 600	3300 ± 600
-NH ₂	8400	5600 ± 800*	9400 ± 800*
-NH-	3100	3100 ± 800	—
-N<	5000	—	—
-NO ₂	1500	1700 ± 200	1700 ± 200
=PO ₄	13000	—	—

* Data from Bondi⁸ corrected to 25°C and $V\delta_p^2$ subtracted. For important steric shielding effects, see his Tables 7.6 and 7.7.

Tabla tomada de (Hansen, 2007)